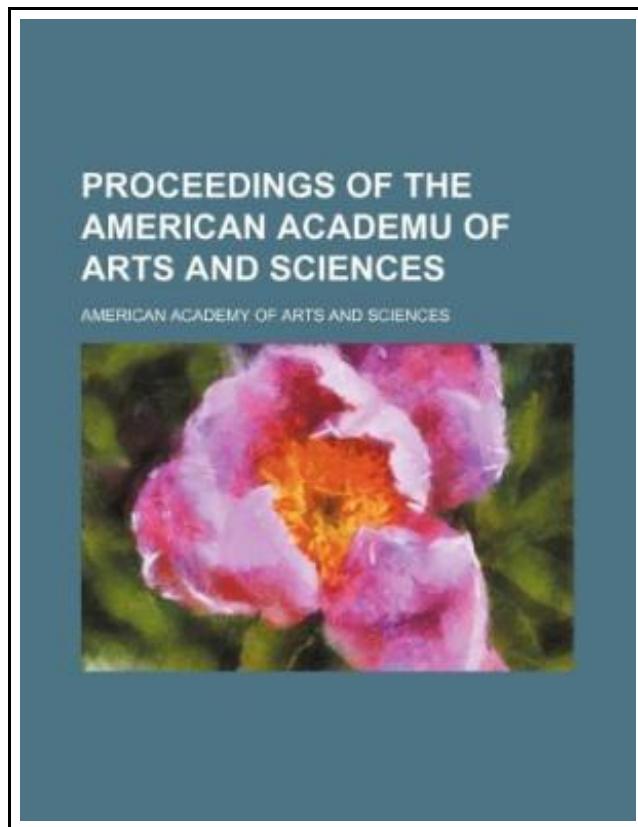


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1910 Excerpt: .measurements of H indicate that its value at 100 is 639.11 mean calories. If this be accepted, the proposed formula for H is $H = 639.11 + 0.3745 (t-100) - 0.000990 (t-100)^2$ mean calories. The last two terms of the formula are the real contribution of this paper, and may still be valid, even if the first term is later found to be wrong. 3. Thiesen s formula for L with recomputed constants is shown to represent satisfactorily all of the reliable values of L, including those in this paper. It is believed to be the safest known means of extrapolating to high temperatures. 4. The literature on the specific heat of superheated steam is systematically discussed and revised in the light of the new values of H - Third paper; Proe. Roy. Soc., 1894, 66, 171; translated in Zeitsch. Phys. Chem., 1894, 15, 106. and of the Joule-Thomson coefficient presented in an earlier paper. In particular the choice of Knoblauch s values of Gt as the foundation of the determination of H in this paper is justified. 5. It is shown that Clausius specific heat of saturated steam passes its maximum without becoming zero or positive, so that the temperature-entropy diagram for steam must be essentially simpler than that for either ether or chloroform. 6. The extrapolation formula for L mentioned in 3 above is made the basis of a determination of the critical density of water by the method of Cailletet and Mathias. The result is...

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